

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property
Organization
International Bureau



(43) International Publication Date
28 July 2005 (28.07.2005)

PCT

(10) International Publication Number
WO 2005/068435 A1

(51) International Patent Classification⁷: C07D 239/42

MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(21) International Application Number:
PCT/CZ2004/000088

(22) International Filing Date:
17 December 2004 (17.12.2004)

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
PV 2004-86 16 January 2004 (16.01.2004) CZ

Declaration under Rule 4.17:

— as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii)) for the following designations AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC, VN, YU, ZA, ZM, ZW, ARIPO patent (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG)

(71) Applicant (for all designated States except US): ZENTIVA, a. s. [CZ/CZ]; U kabelovny 130, Dolni Mecholupy, 102 37 Praha 10 (CZ).

(72) Inventors; and

(75) Inventors/Applicants (for US only): SEBEK, Pavel [CZ/CZ]; Radcina 2/521, 161 00 Praha 6 (CZ). RADL, Stanislav [CZ/CZ]; Pertoldova 3380, 143 00 Praha 2 (CZ). STACH, Jan [CZ/CZ]; Slitrova 2006, 190 00 Praha 9 - Ujezd nad Lesy (CZ).

(74) Agents: JIROTKOVA, Ivana et al.; Rott, Ruzicka & Guttmann, Patent & Trademark & Law Office, Nad Stolou 12, 170 00 Praha 7 (CZ).

(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD,

Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

WO 2005/068435 A1

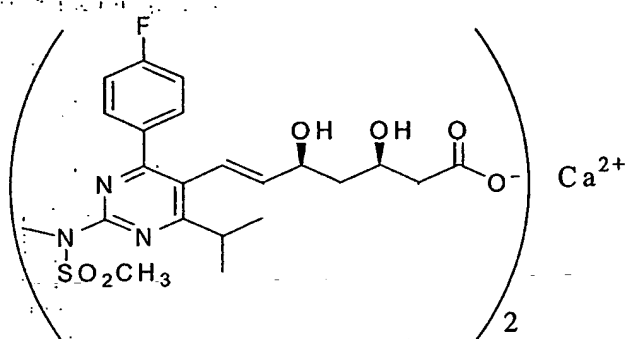
(54) Title: TITLE OF INVENTION: A METHOD OF PREPARATION OF THE HEMI-CALCIUM SALT OF (E)-7-[4-(4-FLUOROPHENYL)-6-ISOPROPYL-2-[METHYL(METHYLSULFONYL)AMINOLPYRIMIDIN-5-YL](3R,5S)-3,5-DIHYDROXY-6-HEPTENOIC ACID

(57) Abstract: A method of preparation of the hemi-calcium salt of rosuvastatin of formula (I) consists in extracting an aqueous solution of the sodium or potassium salt of (E)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3R,5S)-3,5-dihydroxy-6-heptenoic acid, with optional admixture of sodium or potassium hydroxide or other sodium or potassium salts having inorganic anions, with an organic solvent, incompletely miscible with water, selected from the series of R¹COOR², R¹COR² and R¹OII, wherein R¹ and R² independently represent hydrogen or a residue of a C₁-C₁₀ aliphatic hydrocarbon, C₆ aromatic hydrocarbon, C₅ or C₆cyclic hydrocarbon, or a combination of an aliphatic and aromatic or cyclic hydrocarbon, the extract being subsequently shaken with an aqueous solution of an inorganic or C₁-C₃ organic calcium salt, and the product of formula I is further isolated by cooling and/or adding an anti-solvent and filtration, and optionally, is converted into its amorphous form.

A method of preparation of hemi-calcium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid

Technical Field

The invention concerns a new method of preparation of the hemi-calcium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid known under the INN name rosuvastatin, formula I.



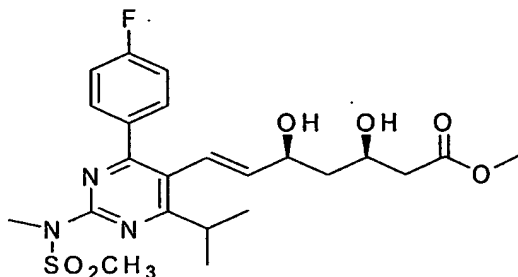
The mentioned medicament is a prominent representative of hypolipidemic and hypocholesteric pharmaceuticals.

Background Art

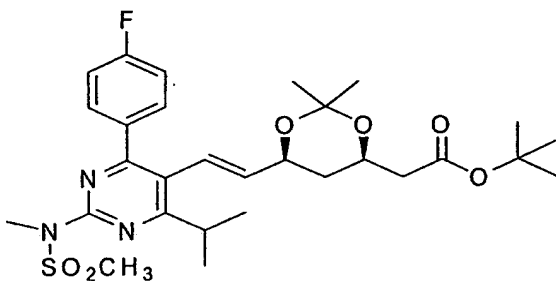
Rosuvastatin is produced according to the published patent (EP 521471) usually from the sodium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid and an appropriate water-soluble calcium salt, preferably from calcium chloride.

The starting sodium salt can be obtained according to the above-mentioned patent from the methyl ester of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid of formula II via hydrolysis with ethanolic sodium hydroxide or lately (according to international patent application WO 00/49014) from *tert*-butyl (*E*)-(6-[2-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl-

(methylsulfonyl)amino]pyrimidin-5-yl]vinyl](4*R*,6*S*)-2,2-dimethyl-[1,3]dioxan-4-yl)-acetate of formula III



II

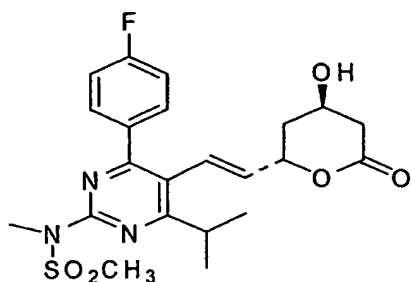


III

5

This intermediate product is first transferred to the corresponding sodium salt by consecutive stirring first with hydrochloric acid and then with sodium hydroxide. The calcium salt is subsequently obtained via addition of calcium chloride to the solution of the sodium salt in water. However, the salt prepared in this way is contaminated with inorganic substances. For example, residual sodium hydroxide reacts with calcium chloride to produce water-insoluble calcium hydroxide. Authors of the new patent application (WO 00/042024) assert that the substance prepared according to patent EP 521471 had an amorphous structure; nevertheless the process of its preparation was difficult to reproduce.

15 According to another patent application (WO 03/016317), the calcium salt can be obtained also via reaction of calcium hydroxide with lactone of formula IV



IV

or other esters of rosuvastatin.

- 5 The objective of this invention is to describe a new, improved method of preparation of the hemi-calcium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)-amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid (rosuvastatin), which would not have the mentioned disadvantages, and also an improved method of preparation of the amorphous form.

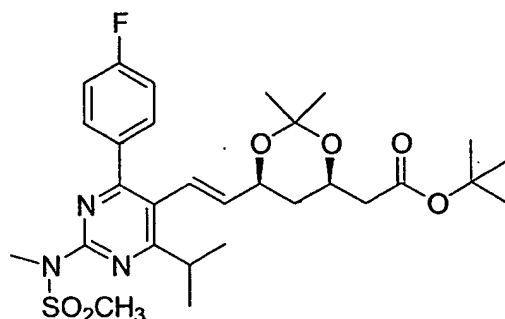
10

Disclosure of Invention

- The subject matter of the invention consists in an improved method of preparation of the hemi-calcium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)-amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid of formula I, wherein an
- 15 aqueous solution of the sodium or potassium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid, with optional admixture of sodium or potassium hydroxide or other sodium or potassium salts having inorganic anions, is extracted with an organic solvent, incompletely miscible with
- 20 water, selected from the series of R^1COOR^2 , R^1COR^2 and R^1OH , wherein R^1 and R^2 independently represent hydrogen or a residue of a C_1 - C_{10} aliphatic hydrocarbon, C_6 aromatic hydrocarbon, C_5 or C_6 cyclic hydrocarbon, or a combination of an aliphatic and aromatic or cyclic hydrocarbon, the extract being subsequently shaken with an aqueous solution of an inorganic or C_1 - C_5 organic calcium salt, and the product of formula I is further isolated by
- 25 cooling and/or adding an anti-solvent and filtration.

The aqueous solution of the sodium or potassium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic

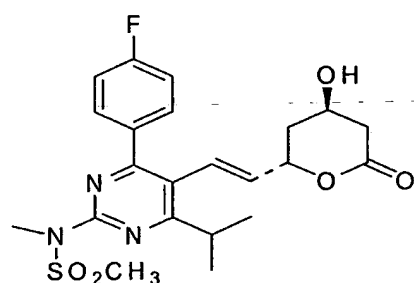
acid is preferably obtained stepwise by acidic hydrolysis and subsequent alkaline hydrolysis of the protected ester of formula III



III

5

or by alkaline opening of the lactone of formula IV



(IV)

- 10 Extraction of the sodium or potassium salt from the aqueous solution is performed with an ester of formula R^1COOR^2 , wherein R^1 and R^2 have the above mentioned meanings, or, even more preferably, extraction is made with ester R^1COOR^2 , wherein R^1 and R^2 are independently hydrogen or a C_1 - C_5 aliphatic residue, preferably with ethyl acetate.
- 15 This whole procedure is based on the surprising finding that the sodium or potassium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid can be quantitatively extracted from the aqueous phase into solvents of the type of esters, ketones or alcohols of formulae R^1COOR^2 , R^1COR^2 or R^1OH , wherein R^1 and R^2 have the above-mentioned meaning. The sodium or potassium
- 20 salt obtained in this way can be quantitatively transferred into the calcium salt by stirring with an aqueous solution of an inorganic or organic calcium salt. Rosuvastatin can be subsequently obtained by evaporation and crystallization.

Another aspect of the invention consists in a new method of preparation of the amorphous form, which is based on dissolving the calcium salt of rosuvastatin in a suitable solvent and adding the same to an anti-solvent, in which rosuvastatin is completely insoluble or little soluble. A solution of the hemi-calcium salt of rosuvastatin in an organic solvent selected from the series of R^1COOR^2 , R^1COR^2 or R^1OH , wherein R^1 and R^2 have the above-mentioned meaning, is added dropwise to an anti-solvent in which rosuvastatin is insoluble, selected from the series including compounds of formulae R^1H and R^1OR^2 , wherein R^1 and R^2 have the above-mentioned meaning, or water.

10 The compound of formula I is dissolved in a solvent preferably selected from the series of $R^{1'}COOR^{2'}$, $R^{1'}COR^{2'}$ or $R^{1'}OH$, wherein $R^{1'}$ and $R^{2'}$ have the above-mentioned meanings, added dropwise to an anti-solvent in which rosuvastatin is insoluble, selected from the series including compounds of formulae $R^{1'}H$, $R^{1'}OR^{2'}$, wherein $R^{1'}$ and $R^{2'}$ have the above-mentioned meanings, or water.

15 The compound of formula I is preferably dissolved in a solution including ketones, particularly acetone, ethyl methyl ketone, isopropyl methyl ketone, alcohols, particularly methanol, ethanol, isopropanol, or butanols, and further esters, particularly of formic acid, acetic acid or propionic acid with methyl, ethyl or propyl alcohol, and the product is
20 precipitated with solvents including heptane, pentane, cyclohexane, toluene, petroleum ether, diethyl ether or water.

Brief Description of Drawings

25 Figure 1 shows the diffraction pattern of an amorphous sample of the hemi-calcium salt of rosuvastatin.

Detailed description of the invention

30 Esters of rosuvastatin or rosuvastatin lactone of formula IV can be hydrolyzed in aqueous tetrahydrofuran with sodium hydroxide and the resulting sodium salt of rosuvastatin can be quantitatively extracted into the organic phase, preferably with ethyl acetate. The sodium salt obtained in this way is converted into the calcium salt by shaking a solution of the sodium salt

in ethyl acetate or another solvent of the above-mentioned type with a water soluble calcium salt, preferably calcium acetate. The residual inorganic contaminants are subsequently removed by washing with demineralized water. Evaporation and crystallization can produce rosuvastatin, which is not contaminated with inorganic substances.

5

According to the original patent EP 521471, the prepared rosuvastatin had an amorphous structure, but the process is not reproducible. The amorphous form has usually different dissolution characteristics and bio-availability than crystalline forms (Konno T.: *Chem. Pharm. Bull.* 1990, 38, 2003). In case of rosuvastatin, which is little soluble in water, it is
10 important to have a reproducible process for obtaining the amorphous form.

In our method, it has turned out that perfectly amorphous rosuvastatin can be obtained by dissolving crystalline or semi-crystalline rosuvastatin in a solvent in which rosuvastatin is soluble under cold conditions or at increased temperatures, selected from the series of
15 R^1COOR^2 , R^1COR^2 or R^1OH , wherein R^1 and R^2 have the above-mentioned meaning, and by adding the resulting solution to an anti-solvent in which rosuvastatin is insoluble, selected from the series of R^1H , R^1OR^2 , wherein R^1 and R^2 have the above-mentioned meaning, or water. The solvents in which rosuvastatin is soluble under cold conditions or at increased temperatures include those solvents in which solubility is higher than 1 g in 50 ml. Mixtures of
20 suitable solvents can be also used. Examples of such preferable solvents include methanol, ethyl methyl ketone or ethyl acetate. The anti-solvents in which rosuvastatin is insoluble include those in which 1g of the substance does not dissolve in 1,000 ml of the solvent under cold conditions. Examples of such solvents include preferably hexane, pentane, diethyl ether or water. A more detailed list of these solvents has been presented above. The diffraction
25 pattern of a perfectly amorphous sample (prepared according to Example 5) is shown in Fig. 1; the measurements were performed on diffractometer SEIFERT 3000 XRD with a graphite monochromator, radiation $CoK\alpha$ ($\lambda = 1.790\text{\AA}$) within the range $2.5 - 40^\circ 2\theta$ with a step 0.03.

The invention is elucidated in more detail in the following examples. The examples, which
30 illustrate preferred alternatives of production of rosuvastatin according to the invention, have a purely illustrative character and do not limit the extent of the invention in any respect. Semi-crystalline rosuvastatin used in Example 5 was obtained according to the original patent EP

521471. Crystalline rosuvastatin used in Examples 6 and 7 was obtained according to WO 00/042024.

Examples

5

Example 1

Tetrahydrofuran (75 ml) is added to lactone IV (5 g, 10.8 mmol). A solution of 40% NaOH (10 ml) is added during 5 minutes to the solution obtained in this way and the formed
10 heterogeneous mixture is vigorously stirred for 17 h and then poured into a separating funnel containing demineralized water (150 ml) and hexane (50 ml). After shaking, the organic layer is separated and the aqueous layer is extracted with a mixture of hexane (40 ml) and tetrahydrofuran (10 ml). After complete separation, the aqueous layer is extracted with ethyl acetate (1 x 40 ml, 3 x 20 ml). The ethyl acetate extract is then gradually shaken 3 times with
15 demineralized water (5 ml), each containing 1 g of calcium acetate in 5 ml of water. The resulting ethyl acetate extract is washed with demineralized water (2 x 5 ml) and, after drying, is concentrated in a vacuum evaporator to a volume of 30 ml and added dropwise to hexane (150 ml) to give, after filtration, 4.5 g of amorphous rosuvastatin.

¹H NMR (DMSO) δ:

20 1.22 (d, J = 7, 6H); 1.41 (m, 1H); 1.61 (m, 1H); 2.18 (dd, J = 3, 2H); 3.43 (m, 1H); 3.45 (s, 3H); 3.57 (s, 3H); 3.83 (m, 1H); 4.25 (m, 1H); 5.56 (dd, J = 7.16, 1H); 6.58 (d, J = 16, 1H); 7.33 (m, 2H); 7.76 (m, 2H)

MS for C₂₂H₂₈FN₃O₆SNa [M + Na]⁺: calculated 504.1; found 503.8.

25 Example 2

Following the procedure described in Example 1 using potassium hydroxide instead of sodium hydroxide for the hydrolysis of the ester, the corresponding potassium salt of rosuvastatin is obtained. The solution is further treated according to the procedure described in Example 1,
30 to provide 4.2 g of amorphous rosuvastatin.

Example 3

Tetrahydrofuran (15 ml) is added to ester III (1 g, 1.7 mmol) and after a clear solution is formed, 10% HCl (4 ml) is added. The mixture is stirred for additional 24 hours at ambient
5 temperature. Then, a solution of 40 % NaOH (2 ml) is added to the solution during 5 min and the formed heterogeneous mixture is vigorously stirred for 17 h and then poured into a separating funnel containing demineralized water (30 ml) and hexane (10 ml). After shaking, the organic layer is separated and the aqueous layer is extracted with a mixture of hexane (8 ml) and tetrahydrofuran (2 ml). After complete separation, the aqueous layer is extracted with
10 ethyl acetate (1 x 20 ml, 3 x 10 ml). Combined ethyl acetate extracts are gradually shaken 3 times with demineralized water (1 ml), each containing 0.2 g of calcium acetate in 1 ml of water. The resulting ethyl acetate solution is washed with demineralized water (2 x 3 ml) and after drying with calcium sulfate, it is evaporated in a vacuum evaporator. After crystallization from acetonitrile and water, 0.7 g of rosuvastatin is obtained.

15

Example 4

Tetrahydrofuran (15 ml) is added to ester II (1 g, 2 mmol) and after complete dissolution, a solution of 40 % NaOH (2 ml) is added to the solution over 5 min and the formed
20 heterogeneous mixture is vigorously stirred for 17 h and then poured in a separating funnel containing demineralized water (30 ml) and hexane (10 ml). After shaking, the organic layer is separated and the aqueous layer is extracted with a mixture of hexane (8 ml) and tetrahydrofuran (2 ml). After complete separation, the aqueous layer is extracted with ethyl acetate (1 x 20 ml, 3 x 10 ml). The ethyl acetate solution is subsequently shaken 3 times with
25 demineralized water (1 ml), each containing 0.2 g of calcium acetate in 1 ml of water. The resulting ethyl acetate solution is washed with demineralized water (2 x 3 ml) and evaporated in a vacuum evaporator. After crystallization from acetonitrile and water, 0.7 g of rosuvastatin is obtained.

30 Example 5

Semi-crystalline rosuvastatin (1 g) is dissolved in ethyl methyl ketone (10 ml) at 40 °C. After being filtered, the resulting solution is added dropwise to pentane (70 ml), while the mixture is

vigorously stirred. After 30 min of stirring, the solution is sucked off and dried in vacuo to give 0.95 g of amorphous rosuvastatin.

Example 6

5

Crystalline rosuvastatin (1.5 g) is dissolved in methanol (10 ml) at 25 °C. After being filtered, the resulting solution is added dropwise to water (150 ml), while the mixture is vigorously stirred at 5 °C. After 30 min of stirring, the solution is sucked off and dried in vacuo to give 1.3 g of amorphous rosuvastatin.

10

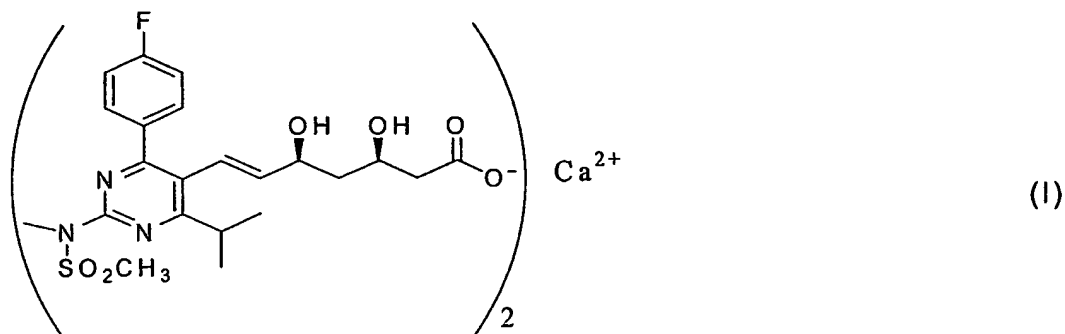
Example 7

Crystalline rosuvastatin (1 g) is dissolved in methanol (10 ml) at 25 °C. After being filtered, the resulting solution is added dropwise to diethyl ether (150 ml) at 25 °C. After 30 min of stirring, the solution is sucked off and dried in vacuo to give 0.7 g of amorphous rosuvastatin.

15

C L A I M S

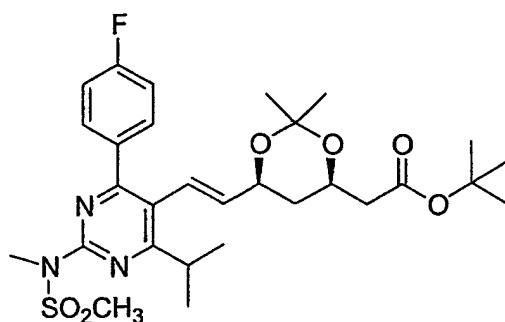
1. A method of preparation of the hemi-calcium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid of
 5 formula I, i.e. rosuvastatin



characterized in that an aqueous solution of the sodium or potassium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid, with optional admixture of sodium or potassium hydroxide or other sodium or potassium salts having inorganic anions, is extracted with an organic solvent, incompletely miscible with water, selected from the series of R^1COOR^2 , R^1COR^2 and R^1OH , wherein R^1 and R^2 independently represent hydrogen or a residue of a C_1 - C_{10} aliphatic
 15 hydrocarbon, C_6 aromatic hydrocarbon, C_5 or C_6 cyclic hydrocarbon, or a combination of an aliphatic and aromatic or cyclic hydrocarbon, the extract being subsequently shaken with an aqueous solution of an inorganic or C_1 - C_5 organic calcium salt, and the product of formula I is further isolated by cooling and/or adding an anti-solvent and filtration, and, optionally, it is converted into its amorphous form.

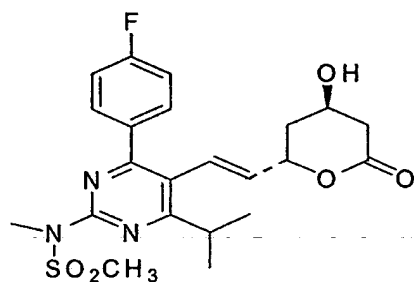
20

2. The method according to claim 1 characterized in that the aqueous solution of the sodium or potassium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid is obtained stepwise by acidic hydrolysis and subsequent alkaline hydrolysis of the protected ester of
 25 formula III



III

or by alkaline opening of the lactone of formula IV



(IV)

5

3. The method according to claim 1 **characterized in** that the extraction of the sodium or potassium salt from the aqueous solution is performed with an ester of formula R^1COOR^2 , wherein R^1 and R^2 are as defined in claim 1.

10

4. The method according to claim 1 **characterized in** that the extraction is performed with ester R^1COOR^2 , wherein R^1 and R^2 are independently hydrogen or a C_1 - C_5 aliphatic residue, preferably with ethyl acetate.

15

5. A method of the preparation of the amorphous form of the hemi-calcium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid of formula I, i.e. rosuvastatin, according to claim 1, **characterized in** that a solution of the hemi-calcium salt of (*E*)-7-[4-(4-fluorophenyl)-6-isopropyl-2-[methyl(methylsulfonyl)amino]pyrimidin-5-yl](3*R*,5*S*)-3,5-dihydroxy-6-heptenoic acid in an organic solvent selected from the series of R^1COOR^2 , R^1COR^2 and R^1OH , wherein R^1 and R^2 are as defined in claim 1, is added dropwise to a solvent in which

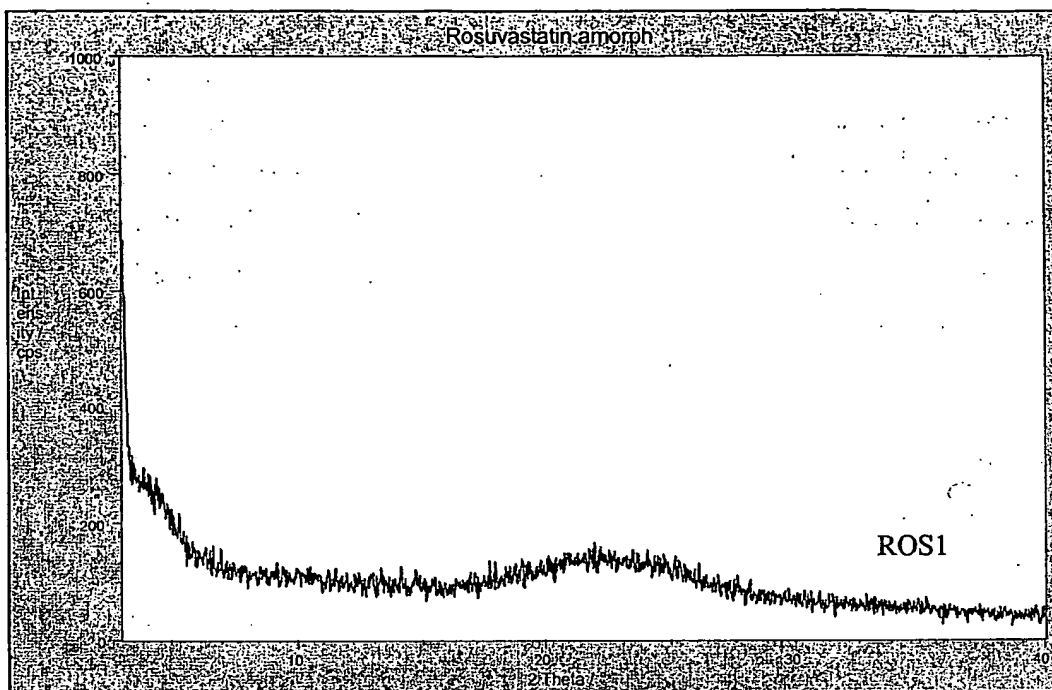
20

rosuvastatin is insoluble, selected from the series including compounds of formulae R^1H and R^1OR^2 , wherein R^1 and R^2 are as defined in claim 1, and water.

6. The method according to claim 5 **characterized in** that the compound of formula I is dissolved in a solvent selected from the series of $R^{1'}COOR^{2'}$, $R^{1'}COR^{2'}$ and $R^{1'}OH$, wherein $R^{1'}$ and $R^{2'}$ are as defined in claim 4, is added dropwise to a solvent in which rosuvastatin is insoluble, selected from the series including compounds of formulae $R^{1'}H$ or $R^{1'}OR^{2'}$, wherein $R^{1'}$ and $R^{2'}$ are as defined in claim 4, and water.
7. The method according to claim 5 **characterized in** that the compound of formula I is dissolved in a solvent including ketones, particularly acetone, ethyl methyl ketone, isopropyl methyl ketone, alcohols, particularly methanol, ethanol, isopropanol, or butanols, further esters, particularly of formic acid, acetic acid or propionic acid with methyl, ethyl or propyl alcohol, and the product is precipitated with solvents including heptane, pentane, cyclohexane, toluene, petroleum ether, diethyl ether or water.

Fig. 1

5



INTERNATIONAL SEARCH REPORT

International Application No.

PCT/CZ2004/000088

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C07D239/42

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 03/016317 A (TEVA PHARMACEUTICAL INDUSTRIES LTD; TEVA PHARMACEUTICALS USA, INC; NID) 27 February 2003 (2003-02-27) cited in the application page 13, line 1 - page 15, line 27 -----	1
A	WO 00/42024 A (ASTRAZENECA UK LIMITED; TAYLOR, NIGEL, PHILLIP) 20 July 2000 (2000-07-20) cited in the application example 1 -----	1
A	EP 0 521 471 A (SHIONOGI SEIYAKU KABUSHIKI KAISHA) 7 January 1993 (1993-01-07) cited in the application example 7 -----	1

☐ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

* Special categories of cited documents:

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *G* document member of the same patent family

Date of the actual completion of the international search

20 April 2005

Date of mailing of the international search report

27/04/2005

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Fanni, S

INTERNATIONAL SEARCH REPORT

International Application No
PCT/CZ2004/000088

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 03016317	A	27-02-2003	US 2002099224 A1	25-07-2002
			CA 2450820 A1	27-02-2003
			CN 1543468 A	03-11-2004
			CZ 20040337 A3	12-01-2005
			EP 1425287 A1	09-06-2004
			HR 20040255 A2	31-08-2004
			JP 2005500382 T	06-01-2005
			NZ 529913 A	24-03-2005
			SK 1402004 A3	03-01-2005
			TR 200302281 T2	21-09-2004
			WO 03016317 A1	27-02-2003
			US 2003114685 A1	19-06-2003
			US 2004176615 A1	09-09-2004
WO 0042024	A	20-07-2000	AT 282027 T	15-11-2004
			AU 762909 B2	10-07-2003
			AU 1882600 A	01-08-2000
			BR 9916786 A	16-10-2001
			CA 2356212 A1	20-07-2000
			CN 1333756 A	30-01-2002
			CZ 20012460 A3	17-10-2001
			DE 69921855 D1	16-12-2004
			EE 200100359 A	16-12-2002
			EP 1144389 A1	17-10-2001
			WO 0042024 A1	20-07-2000
			HU 0104828 A2	29-07-2002
			ID 29432 A	30-08-2001
			JP 2002539078 T	19-11-2002
			NO 20013368 A	05-09-2001
			NZ 512560 A	29-08-2003
			PL 348775 A1	17-06-2002
			RU 2236404 C2	20-09-2004
			SK 9632001 A3	03-12-2001
			TR 200101894 T2	21-12-2001
			US 2004009997 A1	15-01-2004
			US 6589959 B1	08-07-2003
			ZA 200105187 A	23-09-2002
EP 0521471	A	07-01-1993	AT 197149 T	15-11-2000
			CA 2072945 A1	02-01-1993
			CY 2226 A	18-04-2003
			DE 69231530 D1	30-11-2000
			DE 69231530 T2	13-06-2001
			DK 521471 T3	05-02-2001
			EP 0521471 A1	07-01-1993
			ES 2153824 T3	16-03-2001
			GR 3035189 T3	30-04-2001
			HK 1011986 A1	13-07-2001
			HU 220624 B1	28-03-2002
			HU 61531 A2	28-01-1993
			JP 2648897 B2	03-09-1997
			JP 5178841 A	20-07-1993
			KR 9605951 B1	06-05-1996
			LU 91042 A9	24-11-2003
			NL 300125 I1	01-07-2003
			PT 521471 T	30-04-2001
			US RE37314 E1	07-08-2001
			US 5260440 A	09-11-1993

**This Page is Inserted by IFW Indexing and Scanning
Operations and is not part of the Official Record**

BEST AVAILABLE IMAGES

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images include but are not limited to the items checked:

- ☐ **BLACK BORDERS**
- ☐ **IMAGE CUT OFF AT TOP, BOTTOM OR SIDES**
- ☐ **FADED TEXT OR DRAWING**
- ☐ **BLURRED OR ILLEGIBLE TEXT OR DRAWING**
- ☐ **SKEWED/SLANTED IMAGES**
- ☐ **COLOR OR BLACK AND WHITE PHOTOGRAPHS**
- ☐ **GRAY SCALE DOCUMENTS**
- ☒ **LINES OR MARKS ON ORIGINAL DOCUMENT**
- ☐ **REFERENCE(S) OR EXHIBIT(S) SUBMITTED ARE POOR QUALITY**
- ☐ **OTHER:** _____

IMAGES ARE BEST AVAILABLE COPY.

As rescanning these documents will not correct the image problems checked, please do not report these problems to the IFW Image Problem Mailbox.